

THE ASIAN SCHOOL, DEHRADUN
HOLIDAY HOMEWORK FOR WINTER VACATION 2021 FOR CLASS X

ENGLISH :

1. Topic : Pandemic 2019 :

The project should be based on a thorough research on the 'Pandemic' it should include/ discuss –

- a) What it means or refers to.
- b) How, when and where it was identified first.
- c) The initial reaction of the authorities/ people worldwide.
- d) Steps to contain/ curb it.
- e) Its effects :
 - * Psychological effects
 - Lifestyle- (reviving hobbies, interest)
 - * Family ties
 - * Change in food habits

f) Lessons Learnt – Blessings in disguise- importance of spending quality time with the family, importance of physical fitness etc

2. Interview at least four people (from different backgrounds and age groups) and write your analysis based on their response.

Frame at least five questions related to the Pandemic- effect on their lives, relationship, lessons learnt, change in attitude etc.

Instructions : * It is a group activity- each group will have six students who will divide the work amongst themselves

- * One member will be the leader who will coordinate.
- * Questions should be framed and sent to the respective subject teacher for approval.
- * The groups will be made by the teachers and shared soon.
- * The word limits for the project 800-1000 words.
- * Introduction about the project file will be shared later.

HINDI: 1. अनुच्छेद-लेखन

क) नर हो न निराश करो मन को

ख) नैतिक पतन: देश का पतन

2. पत्र-लेखन

क) आपके भाई साहब के यहां एक पुत्र ने जन्म लिया है। पुत्र जन्म के उपलक्ष्य में भोज की व्यवस्था की गई है। आप अपनी सहेली को इस अवसर पर निमंत्रित करते हुए एक पत्र लिखिए।

ख) शैक्षिक सत्र आरम्भ हो चुका है। बाजार में पाठ्य-पुस्तक उपलब्ध नहीं है। इस समस्या का उल्लेख करते हुए किसी दैनिक समाचार पत्र के सम्पादक को एक पत्र लिखिए।

3. विज्ञापन –निम्नलिखित विषयों पर विज्ञापन लिखिए।

क) 'आनन्दा हर्बल चाय' के लिए लगभग 25–50 शब्दों में एक विज्ञापन लिखिए।

ख) पानी शुद्ध करने के यंत्र का 25–50 शब्दों में एक विज्ञापन लिखिए।

MATHEMATICS :

(A):- THE STUDENTS HAVE TO PREPARE A PORTFOLIO/ PROJECT FOR THE TERM-2 , FOR THE INTERNAL ASSESSMENTS IN MATHEMATICS, AS PER THE GUIDELINES ISSUED BY CBSE.

GUIDELINES FOR THE PORTFOLIO/ PROJECT FOR MATHEMATICS

TOPIC FOR THE PORTFOLIO :

HISTORY OF "QUADRATIC EQUATIONS" AND ITS APPLICATION IN REAL LIFE.

KEY-POINTS ABOUT THE TOPIC . (USE IN POINT-B , UNDER THE HEADING INSTRUCTIONS) ***

- MEANING OF THE WORD EQUATION (HISTORY)
- GEOMETRIC APPROACH OF QUADRATIC EQUATIONS.
- CONTRIBUTION OF THE INDIAN MATHEMATICIAN BRAHMAGUPTA IN QUADRATIC EQUATIONS.
- INTERESTING REAL LIFE MATHEMATICAL EXAMPLES IN QUADRATIC EQUATIONS.
- APPLICATION OF THE QUADARTIC FORMULAE.
- NOTE ON USES AND APPLICATION OF QUADRATIC CURVES IN REAL LIFE (e.g. ARCHITECTURE , PHYSICS , CONSTRUCTION , ENGINEERING, SPORTS, NAVIGATION) *** (FOR STUDENTS HELP)

INSTRUCTIONS:

WHAT SHOULD BE INCLUDED IN THE PORTFOLIO .

(A). COVER PAGE IN THE FORMAT GIVEN BELOW.

WELCOME TO MY PORTFOLIO

NAME:
CLASS AND SECTION.....
NAME OF SCHOOL.....
SCHOOL ADDRESS.....
NAME OF SUBJECT TEACHER.....
SESSION.....

PHOTO

SUBJECT: MATHEMATICS

FINISH THE FOLLOWING SENTENCES:-

GOAL SHEET:

- MY STRENGTH IN MATH INCLUDE:
- MY WEAKNESSES IN MATH INCLUDE:
- THE AREAS IN WHICH I HAVE IMPROVED IN MATHS ARE :
.....
.....
.....

(B). NOTE ON WHAT YOU HAVE LEARNT .

C). PORTFOLIO REFLECTION

- 1). DATE OF START.....
- 2). DATE OF COMPLETION.....
- 3). AREAS OF IMPROVEMENT IN PORTFOLIO.....

(D). SELF ASSESSMENT QUESTIONS IN THE PORTFOLIO RELEVANT TO THE TOPIC (MIN. 5 QUESTIONS)

(E). TEACHERS COMMENTS (TO BE DONE BY THE TEACHER).

NAME OF STUDENT.....

TEACHERS SIGNATURE.....

* NOTE:

- MAKE A CREATIVE PORTFOLIO.
- USE PROJECT SHEETS. (MINIMUM 8 SHEETS)
- USE GRAPH SHEETS IF REQUIRED
- COLOR PEN AND PENCIL CAN BE USED
- STICK FILE TO BE USED FOR THE PORTFOLIO

(B):- THE STUDENTS HAVE TO PREPARE MATHS ACTIVITY FILE FOR THE TERM-2 , FOR THE INTERNAL ASSESSMENTS IN MATHEMATICS, AS PER THE GUIDELINES ISSUED BY CBSE.

GUIDELINES AND INSTRUCTIONS FOR PREPARING THE ACTIVITY FILE:

- The activities should be done in a loose ruled project sheets and fixed in a stick file.
- It should contain the following in the order
 - INDEX
 - ACKNOWLEDGEMENT
 - The figures should be drawn with pencil.
 - The headings should be written in black in and the content in blue ink.
 - Use the graph sheets wherever required.
 - Color pencil and pen can be used.
 - Mention NAME OF THE SCHOOL, your NAME, CLASS AND SECTION, SUBJECT and SESSION in the activity file.
 - Prepare a neat and clean creative activity file.

NOTE: The Pdf Of The Activities To Be Completed Will Be Shared In The Class Whataspp Groups By The Subject Teacher.
The Holiday Homework Has To Be Submitted After The Winter Vacations.

PHYSICS : INSTRUCTIONS:

- 1-Homework is to be done neatly in Physics home work notebook.
- 2-Students are supposed to attempt all 30 questions.

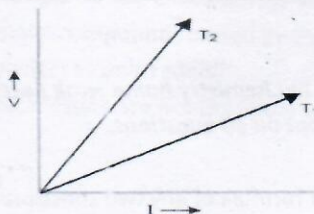
VERY SHORT QUESTIONS

Q1- Write relation between heat energy produced in a conductor when a potential difference V is applied across its terminals and a current I flows through for ' t '

- Q2- State difference between the wire used in the element of an electric heater and in a fuse wire.
- Q3- What is meant by electrical resistivity of a material? Derive its S.I. unit.
- Q4- Write two points of difference between electric energy and electric power.
- Q5- An electrical fuse is rated at 2 A. What is meant by this statement?
- Q6- Write down the relation between the potential difference between two points A and B in a conductor, work done W in moving a unit charge from point B to A and the charge q .
- Q7- Define a solenoid. Compare the magnetic field produced by a solenoid with that of a bar magnet?
- Q8- How can you convert an A.C. into a D.C. generator?
- Q9- Explain why, two magnetic lines of force do not intersect.
- Q10- Give two reasons why different electrical appliances in a domestic circuit are connected in parallel.

SHORT QUESTIONS

- Q11- Distinguish between a solenoid and a bar magnet. Draw the magnetic lines for both.
- Q12- State the right hand thumb rule.
- Q13- What type of core should be put inside a current-carrying solenoid to make an electromagnet?
- Q14- Why is a fuse wire made of a tin-lead alloy and not copper?
- Q15- State the law that gives the relationship between the potential difference (V) across the two ends of a conductor and the current (I) flowing through it.
- Q16- The voltage-current (V - I) graph of a metallic conductor at two different temperatures T_1 and T_2 is shown in figure. At which temperature is the resistance higher



- Q17- A wire of resistivity ' ρ ' is pulled to double its length. What will be its new resistivity?
- Q18- n resistors each of resistance R are connected in parallel in an electric circuit. What is the total effective resistance of the circuit?
- Q19- State Joule's law of heating.
- Q20- List two differences between a voltmeter and ammeter.

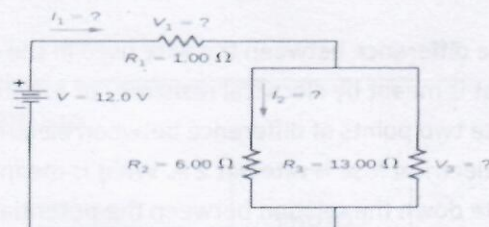
LONG QUESTIONS

- Q21- In the circuit diagram shown, the two resistance wires A and B are of same area of cross-section and same material, but A is longer than B. Which ammeter A_1 and A_2 will indicate higher reading of current? Give reason.
- Q22- A wire of length 3 m and area of cross-section $1.7 \times 10^{-6} \text{ m}^2$ has a resistance $3 \times 10^{-2} \text{ ohm}$.
- What is the formula for resistivity of the wire and what is the unit of it
 - Calculate the resistivity of the wire
- Q23- We have four resistors A, B, C and D of resistance 4 ohm, 8 ohm, 12 ohm and 24 ohm respectively?

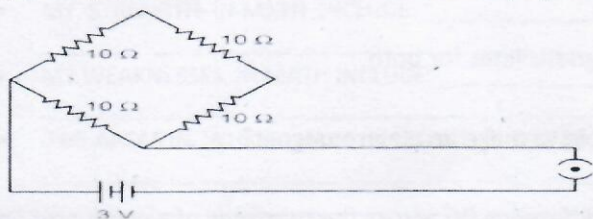
- | | |
|---|--|
| 1 | Lowest resistance which can be obtained by combining these four resistors |
| 2 | highest resistance which can be obtained by combining these four resistors |

- Q24- An electric bulb draws a current of .8 A and works on 250 V on the average 8 hours a day.
- Find the power consumed by the bulb
 - If the electric distribution company charges Rs 5 for 6 KWH, what is the monthly bill for 60 days
- Q25- An electric lamp of 100 Ω , a toaster of resistance 50 Ω , and a water filter of resistance 500 Ω are connected in parallel to a 220 V source. What is the resistance of an electric iron connected to the same source that takes as much current as all three appliances and what is the current through it?

- Q26- a) Find the equivalent resistance of the circuit.
 b) What is the potential drop V_1 across resistor R_1 ?
 c) Find the current I_2 through resistor R_2 .
 d) What power is dissipated by R_2 ?



- Q27- Two lamps rated (100w, 220v) and (60w, 220v) are connected in parallel. Which lamp consumes more energy and which one gives more brightness? Find the resistance of each lamp. (Take supply voltage as 220v).
 Q28- A wire of resistance R is cut into five equal pieces. These five pieces of wire are then connected in parallel. If the resultant resistance of this combination be R then the ratio of resultant to the original will be?
 Q29- Find the current drawn from the battery by the network of four resistors shown in the figure.



- Q30- How much current will an electric bulb draw from 220 V source if the resistance of the bulb is 1200Ω ? If in place of bulb a heater of resistance 100Ω is connected to the sources, calculate the current drawn by

CHEMISTRY:

INSTRUCTIONS:

1-Homework is to be done neatly in Chemistry home work notebook.

2-Students are supposed to attempt all 30 questions.

VERY SHORT QUESTIONS

- Q1- What is the difference in the molecular formula of any two consecutive members of a homologous series of organic compounds?
 Q2- What is meant by a saturated hydrocarbon?
 Q3- What is meant by a functional group in an organic compound?
 Q4- State two characteristic features of carbon which when put together give rise to large number of carbon compounds.
 Q5- How many covalent bonds are there in a molecule of ethane (C_2H_6)?
 Q6- Why do we classify elements?
 Q7- Why do all the elements of the (a) same group have similar properties, (b) same period have different properties?
 Q8- Define Dobereiner triads.
 Q9- Define Newland law of octaves.
 Q10- Define catenation.

SHORT QUESTIONS

- Q11- Why are covalent compounds generally poor conductors of electricity?
 Q12- Write the electron dot structure of ethene molecule (C_2H_4).
 Q13- Write the electron dot structure of ethane molecule (C_2H_6).
 Q14- Out of HCl and CH_3COOH , which one is a weak acid and why? Describe an activity to support your answer.
 Q15- Explain isomerism. State any four characteristics of isomers.
 Q16- Write the name and formula of the 2nd member of the series of carbon compounds whose general formula is C_nH_{2n} .
 Q17- The elements of the third period of the Periodic Table are given below:

Group	I	II	III	IV	V	VI	VII
Period 3	Na	Mg	Al	Si	P	S	Cl

- (a) Which atom is bigger, Na or Mg? Why?
 (6) Identify the most (i) metallic and (ii) non-metallic element in Period 3.
 Q18- Lithium, sodium and potassium form a Dobereiner's triad. The atomic masses of lithium and potassium are 7 and 39 respectively. Predict the atomic mass of sodium.
 Q19- State Mendeleev's periodic law. Write two achievements of Mendeleev's periodic table
 Q20- (a) What is meant by periodicity in properties of elements with reference to the periodic table?
 (b) Why do all the elements of the same group have similar properties?
 (c) How will the tendency to gain electrons change as we go from left to right across a period? Why?

LONG QUESTIONS

Q21- Define homologous series of organic compounds. List its two characteristics. Write the name and formula of the first member of the series of alkenes.

Q22- Why homologous series of carbon compounds are so called? Write chemical formula of two consecutive members of a homologous series and state the part of these compounds that determines their

- (i) physical properties, and (ii) chemical properties.

Q23- What are the hydrocarbons? Write the name and general formula of (i) saturated hydrocarbons, (ii) unsaturated hydrocarbons, and draw the structure of one hydrocarbon of each type.

Q24- Give the electron dot structure and structural formula of first three members of alkene and alkyne families.

Q25- (a) Which two criteria did Mendeleev use to classify the elements in his Periodic Table?

(b) State Mendeleev's periodic law.

(c) Why could no fixed position be given to hydrogen in Mendeleev's Periodic Table?

(d) How and why does the atomic size vary as you go (i) from left to right along a period? (ii) down a group?

Q26- Atoms of seven elements A, B, C, D, E, F and G have a different number of electronic shells but have the same number of electrons in their outermost shells. The elements A and C combine with chlorine to form an acid and common salt respectively. The oxide of element is liquid at room temperature and is a neutral substance, while the oxides of the remaining six elements are basic in nature. Based on the above information, answer the following questions: What could the element A be?

(i) Will elements A to G belong to the same period or same group of the periodic table?

(iii) Write the formula of the compound formed by the reaction of the element A with oxygen,

(iv) Show the formation of the compound by a combination of element C with chlorine with the help of electronic structure.

(v) What would be the ratio of number of combining atoms in a compound formed by the combination of element A with carbon?

(vi) Which one of the given elements is likely to have the smallest atomic radius?

Q27- In the following table, six elements A, B, C, D, E and F (here letters are not the usual symbols of the elements) of the Modern Periodic Table with atomic numbers 3 to 18 are given:

3	4	5	6	7	8	9	10
A					E		G
11	12	13	14	15	16	17	18
B	C		D			F	

(a) Which of these halogen? is (i) a noble gas, (ii) a halogen

(b) If B combines with F, what would be the formula of the compound formed?

(c) Write the electronic configurations of C and E.

Q28- In the following table, are given eight elements A, B, C, D, E, F, G and H (here letters are not the usual symbols of the elements) of the Modern Periodic Table with the atomic numbers of the elements in parenthesis.

Period	Group 1	Group 2
2	A (3)	E (4)
3	B (11)	F (12)
4	C (19)	G (20)
5	D (37)	H (38)

- What is the electronic configuration of F?
- What is the number of valence electrons in the atom of F?
- What is the number of shells in the atom of F?
- Write the size of the atoms of E, F, G and H in decreasing order,
- State whether F is a metal or a non-metal.
- Out of the three elements B, E and F, which one has the biggest atomic size?

Q29- Two elements X and Y belong to group 1 and 2 respectively in the same period of periodic table. Compare them with respect to: periodic table from the left to the right and why?

- the number of valence electrons in their atoms
- their valencies
- metallic character
- the sizes of their atoms
- formulae of their oxides
- formulae of their chlorides

K	L	M
2	8	6

Q30- An element 'E' has following electronic configuration:

- To which group of the periodic table does element 'E' belong?
- To which period of the periodic table does element 'E' belong?
- State the number of valence electrons present in element 'E'.
- State the valency of the element 'E'.

BIOLOGY

INSTRUCTIONS:

1-Homework is to be done neatly in Biology home work notebook.

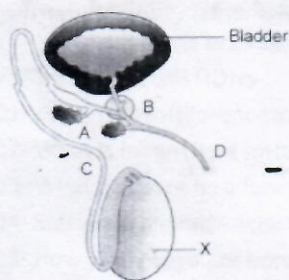
2-Students are supposed to attempt all 30 questions.

VERY SHORT QUESTIONS

- Q1- Name the part of a seed which (i) contains stored food (ii) grows into root, and (iii) grows into shoot.
- Q2- What is the effect of DNA copying which is not perfectly accurate on the reproduction process?
- Q3- Why is it an advantage for the testes to be situated in the scrotal sac outside the main body cavity? Can you think of one disadvantage?
- Q4- In the human body what is the role of: (i) seminal vesicles, and (ii) prostate gland?
- Q5- How does copper-T prevent pregnancy?
- Q6- Name a tiny fresh-water animal which reproduces by the same method as that of yeast. What is this method known as?
- Q7- What step has been taken by UNEP in 1987 to prevent too much damage to the ozone layer?
- Q8- Distinguish between biodegradable and non-biodegradable pollutants.
- Q9- Why has the practice of using 'kulhads' been discontinued?
- Q10- What percentage of the solar energy is trapped and utilized by the plants?

SHORT QUESTIONS

- Q11- Give a reason for each of the following:
- (a) Regeneration is not considered a method of reproduction.
 - (b) Blocking of the fallopian tube causes birth control.
 - (c) Spores generally have a thick wall.
- Q12- In the diagram of the human male reproductive system:



- (a) Label 'X'.
 - (b) Name the hormone produced by 'X'. What is the role of this hormone in human male?
 - (c) Mention the name of substances that are transported by tube C.
- Q13- Illustrate the following with the help of suitable diagrams:
- (i) Regeneration in *Planaria*.
 - (ii) Budding in *Hydra*.
- Q14- Illustrate the following with the help of suitable diagrams:
- (i) Binary Fission in *Amoeba*.
 - (ii) Leaf of *Bryophyllum* with buds.
- Q15-(a) Name the scientist who gave the laws of inheritance.
- (b) Name an animal in which individuals can change sex. What does this indicate?
- Q16- (a) Explain the terms (i) Implantation (ii) Placenta
- (b) What is the average duration of human pregnancy?
- Q17- What are sexually transmitted diseases?

Name four such diseases. Which one of them damages the immune system of human body?

Q18- State two advantages of using disposable paper cups over disposable plastic cups.

Q19- How is ozone formed in the upper atmosphere? Why is the damage of ozone layer a cause of concern to us? State a cause of this damage.

Q20- Explain 'biological magnification' with the help of an example.

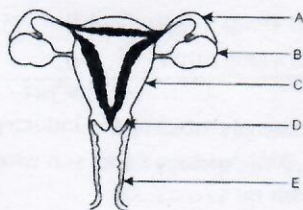
LONG QUESTIONS

- Q21- (a) What are the roles of decomposers in an ecosystem?
- (b) Explain, why a food chain usually cannot have more than three or four steps?
- Q22- In a food chain, 10,000 joules of energy is available to the producer. How much energy will be available to the secondary consumer to transfer it to the tertiary consumer?

Q23- Draw a longitudinal section of a flower and label the following parts:

- (i) part that produces pollen grain.
- (ii) part that transfers male gametes to the female gametes.
- (iii) part that is sticky to trap the pollen grain.
- (iv) part that develops into a fruit.

Q24 - a) Name the parts labelled A, B, C, D and E.



(b) Where do the following functions occur?

- (i) Production of an egg
- (ii) Fertilisation
- (iii) Implantation of zygote.

(c) What happens to the lining of uterus:

- (i) before release of a fertilised egg?
- (ii) if no fertilisation occurs?

Q25- (a) Draw a diagram showing germination of pollen on stigma of a flower.

(b) Label pollen grain, male germ-cells, pollen tube and female germ-cell in the above diagram.

(c) How is zygote formed in a flowering plant?

Q26- A violet colour flower plant denoted by WW is cross bred with that of white colour flower plant denoted by ww.

(a) State the colour of flower you would expect in their F_1 generation plants.

(b) What must be the percentage of white flower plants in F_2 generation if flowers of F_1 plants are self-pollinated?

(c) State the expected ratio of the genotypes WW and Ww in the F_2 progeny.

Q27- If we cross pure-bred tall (dominant) pea plant with pure-bred dwarf (recessive) pea plant we will get pea plants of F_1 generation. If we now self-cross the pea plant of F_2 generation, then we obtain pea plants of F_2 generation.

(a) What do the plants of F_2 generation look like?

(b) State the ratio of tall plants to dwarf plants in F_2 generation.

(c) State the type of plants not found in F_2 generation but appeared in F_2 generation, mentioning the reason for the same.

Q28- Explain the mechanism of sex determination in humans with the help of a flow chart. Which of the two parents, the mother or the father, is responsible for determination of sex of a child?

Q29- In a human, how many chromosomes are present in :

- (a) a brain cell?
- (b) a sperm in the testes ?
- (c) an egg which has just been produced by the ovary ?
- (d) a skin cell ?
- (e) a fertilised egg ?

Q30- Explain how Mendel's experiment shows that traits are inherited independently.

POLITICAL SCIENCE :

Q1 What is the role of an ordinary citizen in reforming the political parties?

Q2 What is the role of opposition party in a democracy?

Q3 Differentiate between National and regional political parties?

Q4 In what way lack of internal democracy is seen in the political parties?

Q5 State the various functions political parties perform in democracy?

Q6 Write a short note on Bahujan Samaj Party (BSP).

Q7 What efforts have been made to reform political parties in India?

Q8 In what way is transparency ensured by a democratic system?

Q9 Explain briefly the outcomes of democracy?

Q10 The economic growth rate in dictatorship is better than that in democratic rule. State reasons for it.

HISTORY:

Q1. Who designed the Swaraj flag? What were the features of this flag? How was it used as a symbol of defiance?

Q2. Why did Mahatma Gandhi feel the need to launch a broad based movement in 1920?

Q3. Explain the role of women in the Civil Disobedience movement?

Q4 How did icon and symbols of India developed the sense of collective belongingness?

Q5 Explain some economic effects of the Non-Cooperation movement.

Q6 What is meant by Satyagraha? How did Gandhiji apply the idea of Satyagraha in our country?

- Q7 Why did political leaders differ sharply over the question of separate electorates?
- Q8 Discuss the Salt March to make clear why it was an effective symbol of resistance against colonialism?
- Q9 Mention three reasons why the rich peasant communities took active participation in the Civil Disobedience movement?
- Q10 Why did Mahatma Gandhi start a nationwide hartal on 6th April against the Rowlatt Act?

GEOGRAPHY :

- Q1. How are Integrated Steel Plants different from Mini Steel Plants? What problems does the industry face? What recent developments have led to a rise in the production capacity?
- Q.2 How do industries pollute the environment?
- Q.3 Discuss the steps to be taken to minimise environment degradation by industry.
- Q.4 In which region are the most of the Jute Mills of India concentrated? Why? What are the challenges faced by the industry?
- Q. 5 Write about the role of Information Technology Industry in modern India. What are Software Technology Parks and where in India are they located?
- Q.6 Write a note on the changing nature of the international trade in the last fifteen years.
- Q.7 'Roadways still have an edge over Railways in India.' Support the statement with arguments.
- Q.8 What is mass communication? What are the different means of mass communication? What is the significance of mass communication in a country like India?
- Q.9 Discuss the importance of tourism in India.
- Q.10 Write a short note on : a) Personal written communication b) Telecom network of India

ECONOMICS:

- Q1 .Modern currency is without any use of its own'; then why is it accepted as a medium of exchange?
- Q2. What is meant by double coincidence of wants?
- Q3. Compare formal sector loans with informal sector of loans regarding interest only.
- Q4. Which country has successfully organized SHGs? Who had initiated the programme?
- Q5. What is collateral? Why do lenders ask for collateral while lending? Explain .
- Q6. What was the reason for putting barriers to foreign trade and foreign investment by the Indian government? Why did it wish to remove these barriers?
- Q27. What are the various ways in which MNCs set up, or control, production in other countries?
- Q28. Differentiate between foreign trade and foreign investment.
- Q29. 'Starting around 1991 some far reaching changes in policy were made in India' . Explain
- Q30. Answer the following questions:(i) Write any two feature of Multinational Corporations.
(ii) Why MNCs are investing in China?
(iii) Why have the MNCs set up their customer care centres in India?

IT:

- Q1. Define the following :
- (a) Primary Key (ii) Foreign Key (iii) Tuple (iv) Attribute (v) Table (vi) Data Concurrency (vii) Data Redundancy
- Q2. Differentiate between DDL and DML commands .Write the commands which come under each category.
- Q3. What are the qualities of successful entrepreneurs?
- Q4. What are the misconceptions about Entrepreneurs?
- Q5. What are the different types of Internet connectivity methods?
- Q6. Define the following (i) Blog (ii) Qumana Application purpose (iii) Instant Messaging (iv) OLTP (v) Packet with reference to data transfer.
- Q7. Explain different kinds of online threat.
- Q8. Define "Accessibility" features in Windows operating system. Briefly explain any five of them.
- Q9. Explain the term accident and emergency.
- Q10.(a) Write the query to create a table with following structure :

Table Product

Field	Data Type
PID	Char(4)
Pname	Varchar(20)
Description	Varchar(40)
Price	Integer

VendorID	VName	DateOfReg	Location
V001	Amul	2020-10-10	Gujrat
V002	Kwality Walls	2001-05-06	Mumbai
V003	Mother Dairy	2002-02-03	Delhi

(b) Considering the following table , write the queries for the questions from (i) to (v)

Table : Vendor

- (i) Write query to display all records
- (ii) Write query to display the records whose location is Delhi
- (iii) Write query to add a new record to the table
- (iv) Write query to modify the record of Vendor ID V003 ; set new value for Vname to Havmor
- (v) Write query to delete all the records from the table

FRENCH :

Q1. Écrivez le reportage sur **le festival de Cannes 2019** le meilleur prix du film, acteurs et actrices etc. avec les images dans votre cahier. (100-150 mots)

Q2. Écrivez **un message** à votre cousin(e) qui est chômage en lui disant de l'annonce vous avez lu récemment dans le journal.

Q3. Vous avez lu récemment un roman Intéressant. Écrivez une lettre à votre ami(e) en racontant l'histoire. (80 mots)

Q4. Faites un projet sur la système éducatif de France et l'Inde

(Comparer la système éducatif de deux Pays)


Remarque:

- Une présentation PowerPoint doit être fait (10 à 12 diapositives).
- La présentation PowerPoint doit être en français.
- Vous devez faire une présentation et vous souhaitez poser des questions.

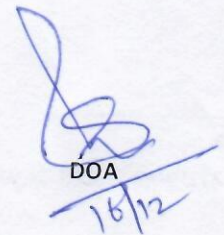
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Classe: Dix

Prenom: _____


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